

A comparative analysis of III, IV, and VII by thin-layer chromatography (TLC) on Al_2O_3 shows that IV and VII lose their basic character. While a system of solvents saturated with dry ammonia is used for the chromatography of III, a low-polarity toluene-chloroform system is used for IV and VII.

EXPERIMENTAL

2,2'-Dipiperidyls III and IIIa. A 50-g (0.3 mole) sample of II with bp 124° (2 mm) and n_D^{20} 1.5412 (bp $256-266^\circ$ [1]) was dissolved in 600 ml of freshly distilled isoamyl alcohol, and 46 g (2 g-atom) of sodium was added in one lump (!) to the cold mixture, after which the mixture was heated to the boiling point on a sand bath. At the end of the reaction, the mixture was cooled, hydrochloric acid was added (pH 5-5.5), and the isoamyl alcohol was removed by steam distillation. The hydrochlorides of the bases were treated with 40% KOH solution until the mixture was strongly alkaline, and the bases were extracted with ether. The ether extract was dried with sodium sulfate, the solvent was removed by distillation, and the residue was vacuum-distilled with collection of the fraction with bp $108-112^\circ$ (1 mm) and n_D^{20} 1.5011. Two substances with R_f 0.12 and 0.27 were detected by TLC on activity II Al_2O_3 with elution by ammonia-saturated CHCl_3 -benzene-methanol (15:18:1.5). The substances were separated by preparative TLC on activity II Al_2O_3 with elution with ammonia-saturated CHCl_3 -benzene-methanol (120:144:12). A 0.5-g sample of the mixture of III and IIIa dissolved in 3 ml of chloroform was applied to a 75-by-35-cm plate. The Al_2O_3 sections containing the substances were collected and washed out with methanol. Removal of the solvent gave chromatographically homogeneous III and IIIa.

Compound III was a colorless liquid with mp 30° (in a sealed capillary, bp $110-111^\circ$ (1 mm), n_D^{20} 1.4993, and R_f 0.12. Found: C 71.4; H 11.8; N 16.8%. $\text{C}_{10}\text{H}_{20}\text{N}_2$. Calculated: C 71.4; H 11.9; N 16.7%. The dihydrochloride was obtained as colorless crystals with mp $299-300^\circ$ (from alcohol). Found: Cl 24.4%. $\text{C}_{10}\text{H}_{20}\text{N}_2 \cdot 2\text{HCl}$. Calculated: Cl 24.5%. The picrate had mp $221-222^\circ$ (from water).

Compound IIIa had mp 44° (from petroleum ether) and R_f 0.27. Found: C 71.3; H 11.8; N 16.8%. $\text{C}_{10}\text{H}_{20}\text{N}_2$. Calculated: C 71.4; H 11.9; N 16.7%. The dihydrochloride was obtained as colorless crystals with mp $310-311^\circ$ (from alcohol). Found: Cl 24.4%. $\text{C}_{10}\text{H}_{20}\text{N}_2 \cdot 2\text{HCl}$. Calculated: Cl 24.5%. The picrate had mp $230-231^\circ$ (from water).

Tricyclo[1.6.7.12]-1,12-diaza-13-tridecanone (IV). A 3.36-g (0.02 mole) sample of III was dissolved in 15 ml of anhydrous toluene, the solution was cooled to -15° , and 1 g (0.01 mole) of phosgene dissolved in 20 ml of toluene was added. The solution was heated on a water bath for 6 h, and the solid product (A) was separated. The toluene was removed by distillation, and the residue was vacuum-distilled to give 1.4 g (71%) of IV as a colorless liquid (the material darkened in air) with bp 126° (1 mm), n_D^{20} 1.5150, and R_f 0.34 [toluene-chloroform-methanol (18:15:1.5)]. Found: C 66.8; H 9.4; N 14.2%. $\text{C}_{11}\text{H}_{18}\text{N}_2\text{O}$. Calculated: C 66.9; H 9.3; N 14.4%. The picrate had mp $175-176^\circ$ (from alcohol). Found: N 16.41%. $\text{C}_{11}\text{H}_{18}\text{N}_2\text{O} \cdot \text{C}_8\text{H}_3\text{N}_3\text{O}_7$. Calculated: 16.5%.

1-Chlorocarbonyl-2,2'-dipiperidyl Hydrochloride (V). A 2-g sample of A in 50 ml of acetone was heated on a water bath for 3 h, solid portion B was separated, and one-third of the solvent was removed by distillation. Petroleum ether (50 ml) was added to the residue to give 0.4 g of colorless crystals with mp $293-294^\circ$ (from acetone) and R_f 0.18 [chloroform-benzene-acetone (15:10:5)]. Found: C 49.4; H 7.2; Cl 29.3; N 15.0%. $\text{C}_{11}\text{H}_{19}\text{ClN}_2\text{O} \cdot \text{HCl}$. Calculated: C 49.4; H 7.5; Cl 29.0; N 14.6%. Product B yielded 0.3 g of the dihydrochloride of III with mp $299-300^\circ$ (from alcohol).

1-Ethoxycarbonyl-2,2'-dipiperidyl (VI). A 2.7-g (0.01 mole) sample of hydrochloride IV in 100 ml of acetone was heated for 5 h with 1.3 g (0.02 mole) of sodium ethoxide. The solvent was removed by distillation to give 1.5 g (86%) of VI with bp $127-128^\circ$ (2 mm), n_D^{20} 1.4988, and R_f 0.52 [chloroform-benzene-methanol (15:18:1.5)]. Found: C 69.7; H 10.6; N 12.4%. $\text{C}_{13}\text{H}_{24}\text{N}_2\text{O}$. Calculated: C 69.7; H 10.7; N 12.5%. The hydrochloride had mp $247-248^\circ$ (from alcohol). Found: Cl 12.7%. $\text{C}_{13}\text{H}_{24}\text{N}_2\text{O} \cdot \text{HCl}$. Calculated: Cl 12.8%. The picrate had mp $168-169^\circ$ (from ether). Reaction of 0.5 g (3 mmole) of III in 10 ml of toluene with 0.32 g (3.5 mmole) of ethyl chlorocarbonate gave 0.1 g (30%) of VI with R_f 0.52. The picrate had mp $168-169^\circ$ (from ether).

1,1'-Diethoxycarbonyl-2,2'-dipiperidyl (VII). A solution of 2.5 g (0.01 mole) of VI in 30 ml of anhydrous ether was mixed with cooling with 0.54 g (0.005 mole) of ethyl chlorocarbonate. The ether solution was separated, and the solvent was removed by distillation to give prisms of VII with mp 44° (from petroleum ether at -10°). Found: C 68.5; H 10.1; N 10.2%. $\text{C}_{16}\text{H}_{28}\text{N}_2\text{O}_2$. Calculated: C 68.6; H 10.0; N 10.0%.

LITERATURE CITED

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